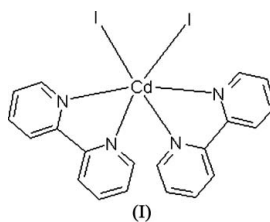


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of ChinaCorrespondence e-mail:  
guohongxu919@yahoo.com.cn**Key indicators**Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.010$  Å  
 $R$  factor = 0.036  
 $wR$  factor = 0.111  
Data-to-parameter ratio = 19.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.***cis*-Bis(2,2'-bipyridine)diiodocadmium(II)**The title compound,  $[\text{CdI}_2(\text{C}_{10}\text{H}_8\text{N}_2)_2]$ , synthesized by hydro-  
thermal methods from  $\text{CdI}_2$  and 2,2'-bipyridine, has the Cd  
atom coordinated by two I and four N atoms from two 2,2'-  
bipyridine ligands in a distorted octahedral geometry.Received 24 April 2006  
Accepted 30 April 2006**Comment**The chemistry of novel metal–organic hybrid coordination  
complexes has been the subject of intensive research in recent  
years owing to their interesting topologies and unexpected  
properties for potential applications (Li *et al.*, 2003; Hammond  
*et al.*, 1999). Recently, there has been increasing interest in  
cadmium–halogen compounds because of their applications in  
molecular materials (Strasdeit *et al.*, 1988; Liu *et al.*, 2002;  
Zhou *et al.*, 2003). In this communication, we have introduced  
2,2'-bipyridine (bpy) as a terminal ligand which favors crystal  
growth of the product. Through a mild-temperature hydro-  
thermal process, we have successfully synthesized the title  
crystalline iodo-coordinated Cd complex,  $[\text{CdI}_2(\text{bpy})_2]$ , (I).The molecular structure of (I) is shown in Fig. 1. The  
compound is a mononuclear complex in which the Cd atom is  
coordinated by two iodide anions and four N atoms from two  
bpy ligands in a distorted octahedral geometry. As shown in  
Table 1, The Cd–I and Cd–N bond lengths are in the  
expected ranges.**Experimental**The hydrothermal reaction of cadmium diiodide (0.095 g, 0.26 mmol),  
2,2'-bipyridine (0.086 g, 0.55 mmol) and water (15.0 ml) was carried  
out at 433 K for 4 d. After cooling to room temperature at a rate of  
 $5 \text{ K h}^{-1}$ , block-shaped brown crystals of (I) suitable for X-ray analysis  
were obtained.*Crystal data* $[\text{CdI}_2(\text{C}_{10}\text{H}_8\text{N}_2)_2]$   
 $M_r = 678.57$   
Monoclinic,  $C2/c$   
 $a = 16.6260$  (10) Å  
 $b = 15.7192$  (9) Å  
 $c = 17.8778$  (13) Å  
 $\beta = 109.642$  (2)°  
 $V = 4400.4$  (5) Å<sup>3</sup> $Z = 8$   
 $D_x = 2.049 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation  
 $\mu = 3.81 \text{ mm}^{-1}$   
 $T = 293$  (2) K  
Block, brown  
 $0.40 \times 0.20 \times 0.12 \text{ mm}$

## Data collection

Rigaku Weissenberg IP diffractometer  
 $\omega$  scans  
 Absorption correction:  $\psi$  scan (TEXRAY; Molecular Structure Corporation, 1999)

$T_{\min} = 0.713$ ,  $T_{\max} = 0.998$   
 (expected range = 0.452–0.633)  
 18349 measured reflections  
 4660 independent reflections  
 3343 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.054$   
 $\theta_{\text{max}} = 27.5^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.111$   
 $S = 1.01$   
 4660 reflections  
 239 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0546P)^2 + 2.0186P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 1.18 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.70 \text{ e } \text{Å}^{-3}$

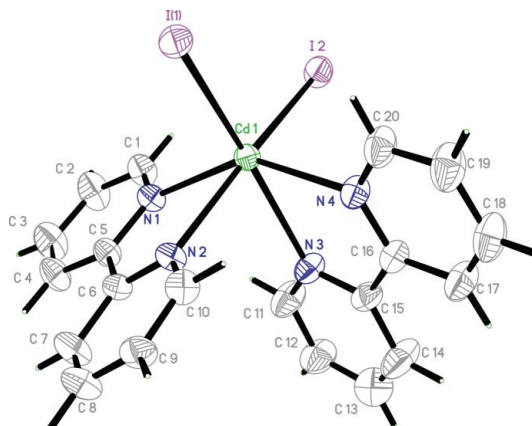
**Table 1**

Selected geometric parameters ( $\text{Å}$ ,  $^\circ$ ).

Cd1–N1	2.377 (4)	Cd1–N3	2.475 (4)
Cd1–N4	2.380 (4)	Cd1–I2	2.8274 (5)
Cd1–N2	2.445 (4)	Cd1–I1	2.8490 (5)
N1–Cd1–N4	146.84 (15)	N2–Cd1–I2	165.87 (10)
N1–Cd1–N2	68.39 (14)	N3–Cd1–I2	92.25 (11)
N4–Cd1–N2	88.50 (15)	N1–Cd1–I1	103.85 (10)
N1–Cd1–N3	84.71 (15)	N4–Cd1–I1	98.67 (10)
N4–Cd1–N3	67.37 (15)	N2–Cd1–I1	88.33 (9)
N2–Cd1–N3	79.56 (15)	N3–Cd1–I1	161.53 (11)
N1–Cd1–I2	99.61 (9)	I2–Cd1–I1	102.208 (17)
N4–Cd1–I2	99.02 (10)		

All H atoms were placed at calculated positions and refined with isotropic displacement parameters using a riding model [ $C-H = 0.93 \text{ Å}$  and  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ ]. Atom C13 was refined with isotropic displacement parameters because of problems with anisotropic refinement. The highest density peak is located  $1.18 \text{ Å}$  from atom I1.

Data collection: TEXRAY (Molecular Structure Corporation, 1999); cell refinement: TEXRAY; data reduction: TEXSAN (Molecular Structure Corporation, 1999); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine



**Figure 1**

View of the structure of the title compound, showing the atom-labelling scheme and displacement ellipsoids drawn at the 30% probability level.

structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEX (McArdle, 1995); software used to prepare material for publication: SHELXL97.

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